

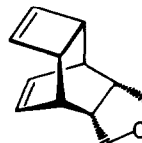
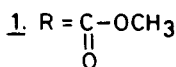
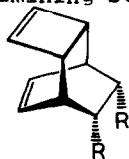
STEREOCHEMISTRY OF BENZENESULPHENYL CHLORIDE ADDITION TO A STRAINED CYCLOBUTENE

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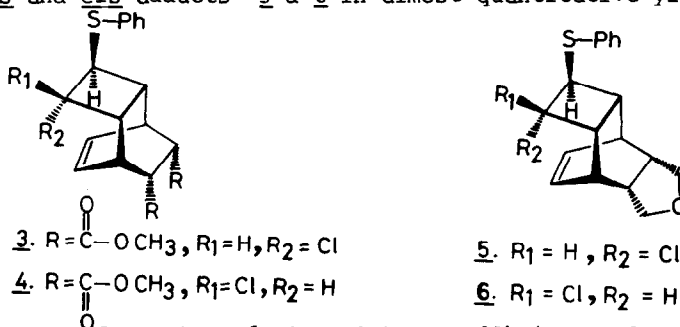
Electrophilic additions of sulphenyl chlorides to olefins have been extensively studied and are considered to be highly stereospecific trans-additions proceeding in two steps via episulphonium (thiiranium) ion intermediates.¹ Simple olefins like cyclohexene^{2a} as well as strained olefins e.g., norbornene,^{2b,3} bicyclo [2.2.1] hex-2-ene^{2c} and acenaphthene,^{2d} with marked propensity for cis additions,⁴ conform to the above mechanism and furnish exclusive trans products with arylsulphenyl chloride. In the case of norbornyl system, this two step addition via episulphonium ion intermediate has been used to arrive at significant mechanistic and stereochemical conclusions.³ This communication reports the first nonstereospecific addition of benzenesulphenyl chloride (PhSCl) to the strained cyclobutene double bond of tricyclo [4.2.2.0^{2,5}] deca-3,7-diene derivatives 1 & 2, which seems to discount the involvement of the bridged episulphonium ion^{1,2} in the product determining step of the reaction.



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Addition of PhSCl to a methylene chloride solution (-20°) of diester 1 containing small amount of suspended calcium carbonate and separation (silica gel) yielded a 3:1 mixture of trans and cis adducts⁵ 3 & 4 in quantitative yield. The stereochemistry of 3 & 4 has been deduced from their pmr spectra.⁶ The olefinic protons at C₇ and C₈ appeared as a sharp triplet ($\delta 6.55$) in the cis-adduct 4 due to symmetrical endo substitution at C₃ and C₄ and as a diffused multiplet ($\delta 6.46$) in the trans adduct 3 due to unsymmetrical endo substitution. Furthermore, the exo proton ($\delta 4.42$) at C₃ in trans adduct 3 exhibited significant shielding

due to the cis vicinal phenylthio group³ as compared to the C₃ endo proton (δ 4.9) in 4. Similarly, the addition of PhSCl to the tetracyclic ether 2 furnished a 3:1 mixture of trans and cis adducts⁵ 5 & 6 in almost quantitative yield:



The simultaneous formation of cis and trans addition products from 1 & 2 with PhSCl can be explained through a free carbonium ion intermediate but is not compatible with the intermediacy¹ of an episulphonium ion. Further support for this contention is derived from the results⁷ of addition of IN₃ and Hg(OAc)₂ to 1 & 2, which proceed via the corresponding three membered intermediates and furnish exclusive cis-addition products under the dominant influence of twist strain⁴ factors.

REFERENCES

- W.H. Mueller, *Angew. Chem. Int. Ed.*, **8**, 482 (1969).
- (a) S.J. Cristol, R.P. Arganbright, G.D. Brindell and R.M. Heitz, *J. Amer. Chem. Soc.*, **79**, 6035 (1957); (b) W.H. Mueller and P.E. Butler, *J. Amer. Chem. Soc.*, **90**, 2075 (1968); (c) F.T. Bond, *J. Amer. Chem. Soc.*, **90**, 5326 (1968); (d) W.H. Mueller and P.E. Butler, *J. Amer. Chem. Soc.*, **88**, 2866 (1966).
- H.C. Brown and K.T. Liu, *J. Amer. Chem. Soc.*, **92**, 3502 (1970).
- T.G. Traylor, *Acc. Chem. Res.*, **2**, 152 (1969).
- Compound 3: C₂₀H₂₁O₄SCl; mp 119-20°; ir(KBr): 1740, 1210, 740 cm⁻¹; pmr(CDCl₃): δ 7.25(5H, s, phenyl), 6.46(2H, m, olefinic), 4.42(1H, t, H-C-Cl), 3.7(1H, t, H-C-S-Ph), 3.56(6H, s, -COOCH₃). Compound 4: C₂₀H₂₁O₄SCl; mp 201-3°; ir(KBr): 1740, 1215, 755 cm⁻¹; pmr(CDCl₃): δ 7.5(5H, s, phenyl), 6.55(2H, t, olefinic), 4.9(1H, t, H-C-Cl), 3.6(6H, s, COOCH₃). Compound 5: C₁₈H₁₉OSCl; mp 143-4°; ir(KBr): 3050, 1575, 740 cm⁻¹; pmr(CDCl₃): δ 7.18(5H, m, phenyl), 6.39(2H, m, olefinic), 4.5(1H, t, H-C-Cl), 3.2-4.1(5H, m, H-C-S-Ph & ether). Compound 6: C₁₈H₁₉OSCl; mp 152-3°; ir(KBr): 3060, 1520, 1050, 745 cm⁻¹; pmr(CDCl₃): δ 7.5(5H, s, phenyl), 6.35(2H, t, olefinic), 4.88(1H, t, H-C-Cl), 3.1-4(5H, m, H-C-S-Ph and ether).
- For the configurational assignment of tricyclo[4.2.2.0^{2,5}]dec-7-enes, see, J.P. Snyder and D.G. Farnum, *J. Org. Chem.*, **31**, 1699 (1966).
- G. Mehta, P.K. Dutta and P.N. Pandey, *Tetrahedron Letters*, 445 (1975); G. Mehta and P.N. Pandey, manuscript submitted for publication. Also see, T. Sasaki, K. Kanematsu, A. Kondo and Y. Nishitani, *J. Org. Chem.*, **39**, 3569 (1974).